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Key indicators

Single-crystal X-ray study T = 293 KMean $\sigma(C-C) = 0.003 \text{ Å}$ R factor = 0.048 wR factor = 0.148 Data-to-parameter ratio = 17.7

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. The title compound, $C_{10}H_{14}O_4$, is used to cure coughs and clear up phlegm and it is also used as an intermediate in the synthesis of other medicinal products. The entire molecule, except for atoms C10 and O4, is essentially planar (to within 0.001 Å).

3-(2-Methoxyphenoxy)propane-1,2-diol

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Comment

The title compound, (I), also known as guaiphenesin, is an ingredient usually found in cold preparations. In the year 1530, it was first extracted from guaiacum and used to treat rheumatism. Over 20 years ago, it was synthesized, named guaiphenesin, and pressed into tablets (Starlanyl, 2001). To the best of our knowledge [using Chemical Abstracts and the Cambridge Structural Database (Allen, 2002)], the single-crystal structure has not been reported previously.



The molecular structure of (I) is shown in Fig. 1. It can be seen that the entire molecule, except for atoms C10 and O4



Figure 1

ORTEPII (Johnson, 1976) view of the title compound, (I), showing the atom-labeling scheme. Displacement ellipsoids are drawn at the 30% probability level.

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and the H atoms, is essentially planar (to within 0.001 Å). The crystal packing projected on to the *ac* face is shown in Fig. 2. There are two kinds of intermolecular hydrogen bonds between molecules (Table 1). The hydrogen bond O3– $H3\cdots O4$ is approximately parallel to the *ac* face, while O4– $H4\cdots O3$ is approximately perpendicular to the *ac* face.

Experimental

The title compound was provided by Tianjin Zhongxin Pharmaceutical Co. Ltd. A saturated solution of guaiphenesin in ethanol was prepared at 338–343 K and then a small quantity of seeds was added when the temperature fell to 308 K. After a long time, a large quantity of white crystals was obtained by recrystallization. The product was characterized by NMR, IR and elemental analyses, and its purity was 99%. The melting point determined by DSC (differential scanning calorimetry) is 355.4 K. Colorless block-shaped single crystals suitable for X-ray diffraction were obtained by adding a small quantity of seeds to a room-temperature solution of the above product and placing it in a refrigerator for 3 d.

Crystal data

 $\begin{array}{l} C_{10}H_{14}O_4 \\ M_r = 198.21 \\ \text{Orthorhombic, } P2_12_12_1 \\ a = 4.9836 \ (10) \text{ Å} \\ b = 7.6562 \ (15) \text{ Å} \\ c = 25.698 \ (5) \text{ Å} \\ V = 980.5 \ (3) \text{ Å}^3 \\ Z = 4 \end{array}$

 $D_x = 1.343 \text{ Mg m}^{-3}$ Mo $K\alpha$ radiation Cell parameters from 9593 reflections $\theta = 3.1-27.5^{\circ}$ $\mu = 0.10 \text{ mm}^{-1}$ T = 293 (2) K Block, colorless

$0.38\,\times\,0.20\,\times\,0.09$ mm

Data collection

Rigaku R-AXIS RAPID IP areadetector diffractometer ω scans Absorption correction: multi-scan (ABSCOR; Higashi, 1995) $T_{min} = 0.962, T_{max} = 0.991$ 9631 measured reflections *Refinement* Refinement on F^2

Refinement on $F^ R[F^2 > 2\sigma(F^2)] = 0.048$ $wR(F^2) = 0.148$ S = 1.002248 reflections 127 parameters H-atom parameters constrained 1862 reflections with *I* > 2*σ*(*I*) $R_{int} = 0.092$ $θ_{max} = 27.5^{\circ}$ h = -6 → 6 k = -9 → 9 l = -33 → 32 $w = 1/[σ^{2}(F_{o}^{2}) + (0.0903P)^{2}]$

2248 independent reflections

 $\begin{array}{l} & = 1/[0^{-}(T_{0}^{-}) + (0.5051^{-})] \\ & + 0.0695P] \\ \text{where } P = (F_{0}^{-2} + 2F_{c}^{-2})/3 \\ (\Delta/\sigma)_{\text{max}} = 0.001 \\ \Delta\rho_{\text{max}} = 0.31 \text{ e } \text{ Å}^{-3} \\ \Delta\rho_{\text{min}} = -0.42 \text{ e } \text{ Å}^{-3} \end{array}$

Table 1 Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$O3-H3B\cdots O4^{i}$ $O4-H4C\cdots O3^{ii}$	0.82 0.82	1.96 1.99	2.744 (2) 2.729 (2)	161 149
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Symmetry codes: (i) -x + 1, $y - \frac{1}{2}$, $-z + \frac{1}{2}$; (ii) x + 1, y, z.

H atoms were placed in calculated positions and constrained to ride on their parent atoms, with C–H = 0.93–0.98 Å and $U_{\rm iso}({\rm H})$ = $1.2U_{\rm eq}({\rm C})$. In the absence of significant anomalous dispersion effects, Friedel equivalents were merged prior to the final refinements, and the absolute configuration was assigned to correspond with the known chiral centers of the precursor molecule, which remained unchanged during the synthesis of the title compound.

Data collection: *RAPID-AUTO* (Rigaku, 2001); cell refinement: *RAPID-AUTO*; data reduction: *RAPID-AUTO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPII* (Johnson, 1976); software used to prepare material for publication: *SHELXL97*.

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